

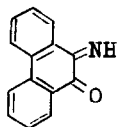
OXAZAPHOSPHOLES. A NEW CLASS OF COMPOUNDS
CONTAINING PHOSPHORUS

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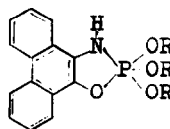
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As far as we aware, compounds of the type mentioned in the title have not hitherto been described. Trialkyl phosphites (distilled over Na), namely, trimethyl-, triethyl-, and triisopropyl phosphites react with phenanthrenequinonemonoimine¹(I), in boiling benzene, to give the first reported 2,2,2,3-tetrahydro-2,2,2-trialkoxyphenanthro(9,10-d)-1,3,2-oxazaphospholes (II).



I



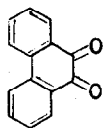
IIa, R=CH₃; b, R=C₂H₅; c, R=C₃H₇-i

The adducts are colourless crystalline substances with sharp m.ps. and are stable² only for few days. Correct combustion values are obtained for all the new compounds (IIa-c), and their molecular weights (the ebullioscopic method in benzene) correspond to the monomeric formulae(cfII).

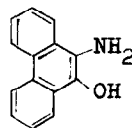
The IR and ¹H n.m.r. spectra³ of adducts IIa-c are compatible with the oxazaphosphole structure. The IR spectrum of IIa in KBr, for example, shows bands at 3200 cm⁻¹ (NH); 1605 cm⁻¹ and 1630 cm⁻¹ (aromatic bands); 1050 cm⁻¹ (P-OCH₃)⁴. No absorption bands are recorded at 1587-1570 cm⁻¹ (enolate)⁵ or at 1350-1250 cm⁻¹ (phosphoryl)⁴ and the region near 1680 cm⁻¹ characteristic of the C=O vibration of the parent compound I was free of absorption. The ¹H n.m.r. spectrum of IIa (CDCl₃, 7) shows the following assignments: multiplet at 2.35 (8 aromatic protons), a doublet at 6.14 (J_{HP} = 11.4 Hz, 9 methoxyl protons).

That the adducts (IIa-c) do not react with diazomethane is also in favour of the cyclic formulae and the regeneration of the starting quinone I upon the pyrolysis of IIa is in agreement with what is known regarding the facile elimination of phosphorus from cyclic compounds⁶.

Treatment of adduct IIa with aqueous 10% sodium hydroxide yields phenanthrenequinone (III). It is very probable that this reaction gives first 9-amino-10-phenanthrol (IV), which is further oxidised by the air to phenanthrenequinone⁷ (III). Oxidation of IIa with chromium trioxide in glacial acetic acid also results in the formation of III.



III



IV

REFERENCES

1. R. Anschutz and G. Schultz, Liebigs Ann., 196, 51 (1879).
2. All manipulations were carried out under N_2 . Moisture and O_2 must be avoided in making and handling the adducts.
3. The IR spectra were recorded with a Carl Zeiss Infracord Spectrophotometer Model "UR 10" and the 1H n.m.r. spectra were run on Varian A 60 Spectrometer, using TMS as internal standard.
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